





Yelin Hu^{1, 2}, Florent Boudoire^{2,3}, Iris Hermann-Geppert^{4, 5}, Peter Bogdanoff⁶, George Tsekouras¹, Bongjin S. Mun⁷, Giuseppino Fortunato⁸, Michael Graetzel², Artur Braun¹



¹ Empa – Swiss Federal Laboratories for Materials Science & Technology, CH-8600 Dübendorf, Switzerland

³ Department of Chemistry, University of Basel, Spitalstr. 51, CH-4056 Basel, Switzerland

⁵ Institute for Materials Technology, Helmut-Schmidt University, D-22043 Hamburg, Germany

⁷ GIST – Gwangju Institute of Science & Technology, Gwangju, Korea

² Laboratory for Photonics and Interfaces, Ecole Polytechnique Federal de Lausanne, CH-1015 Lausanne, Switzerland

⁴ Institute for Materials Research, Sustainable Energy Technology, Helmholtz-Zentrum Geesthacht, D-21502 Geesthacht, Germany

⁶ Institute for Solar Fuels, Helmholtz-Zentrum Berlin für Materialien und Energie, D-14109 Berlin, Germany

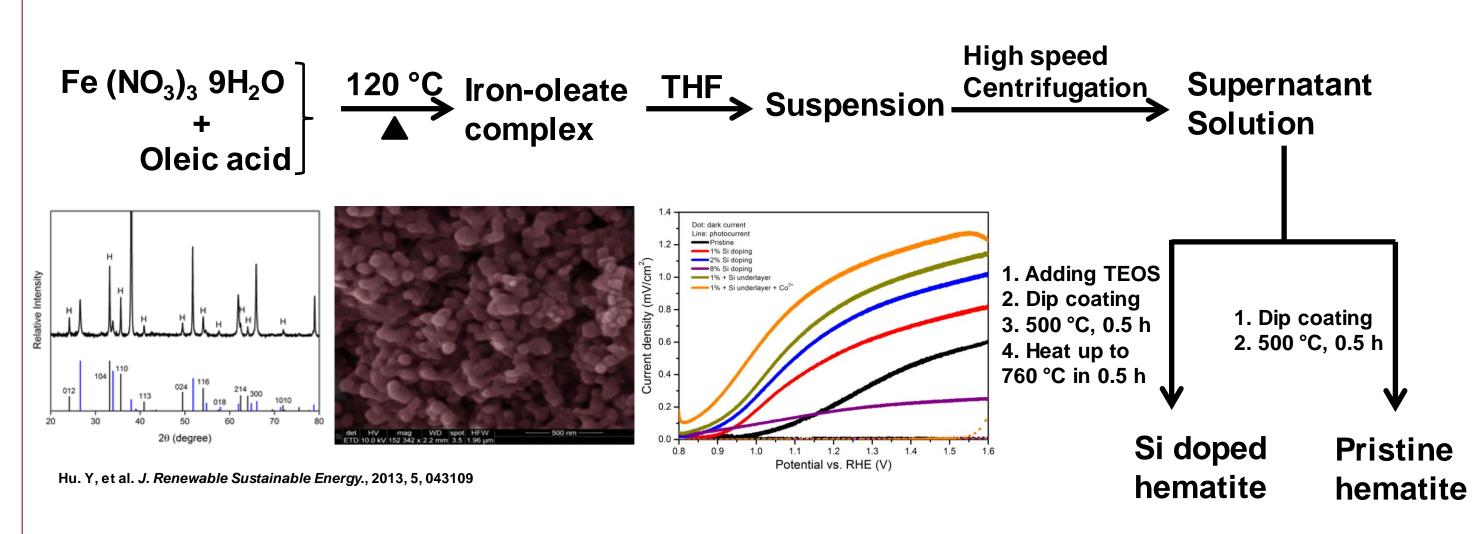
⁸ Protection and Physiology, Empa, Swiss Federal Laboratories for Materials Science and Technology, CH-9014 St. Gallen, Switzerland

Introduction

Hematite (α-Fe₂O₃) is a prospective photoanode material for the oxygen evolution reaction upon water splitting. The surface states of hematite have been under scrutiny for several decades. However, their origin and influence on the photoelectrochemical performance is still poorly understood. In the present study, hematite films were prepared by dip-coating fluorine-doped tin oxide coated glass substrate followed by surface modification via oxygen plasma treatment. O 1s core level X-ray photoelectron spectra and resonant valence band photoemission at Fe 3p edge suggested the filling of oxygen vacancies and oxidation of Fe²⁺ upon oxygen plasma treatment. Electrochemical impedance spectroscopy was employed to determine the degree of charging of surface states. An existence of strong correlation between oxygen vacancies, surface states and photocurrent density was demonstrated.

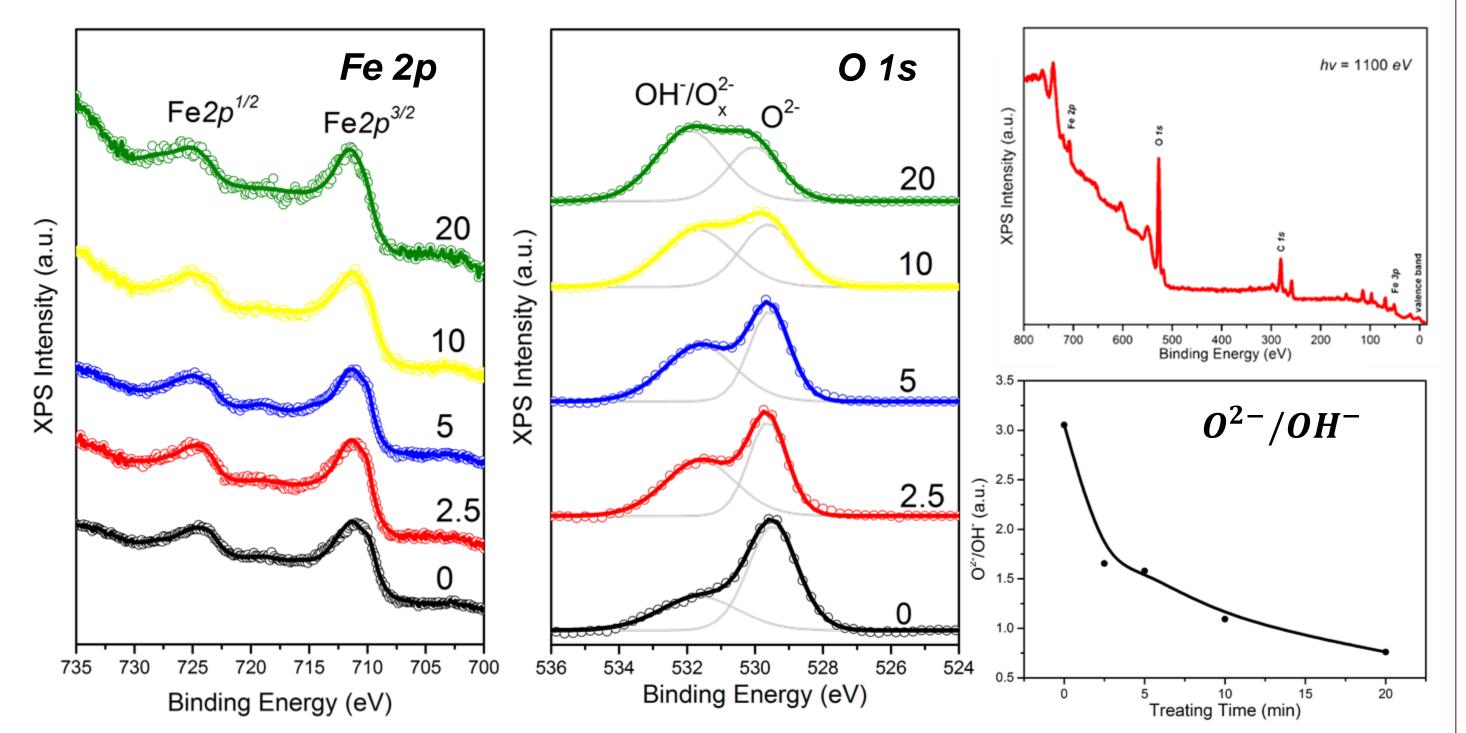


Dip coating method for hematite preparation

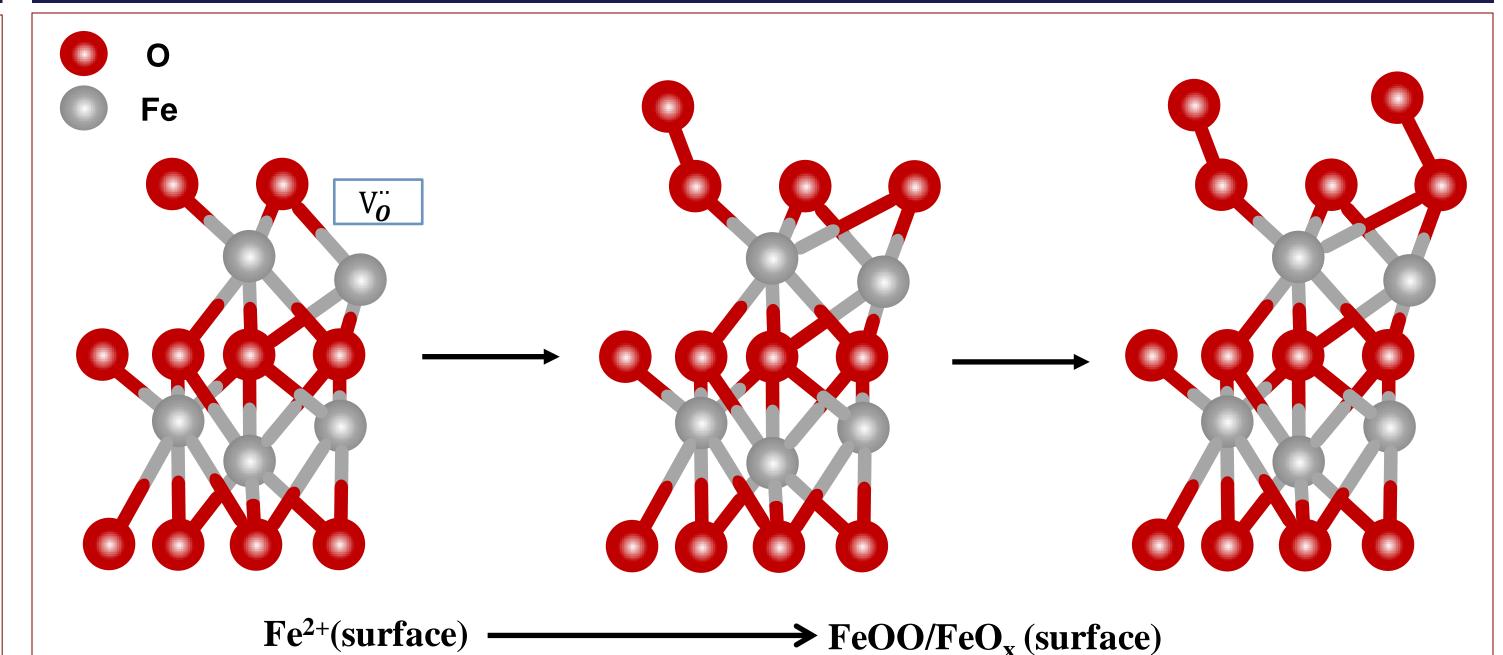


Photocurrent density over 1 mA/cm² at 1.23 V_{RHE} achieved for Si doped hematite sample
Undoped hematite samples were post-treated *via* oxygen plasma for this study

XPS analysis on plasma treated hematite samples



Surface during O₂ plasma treatment

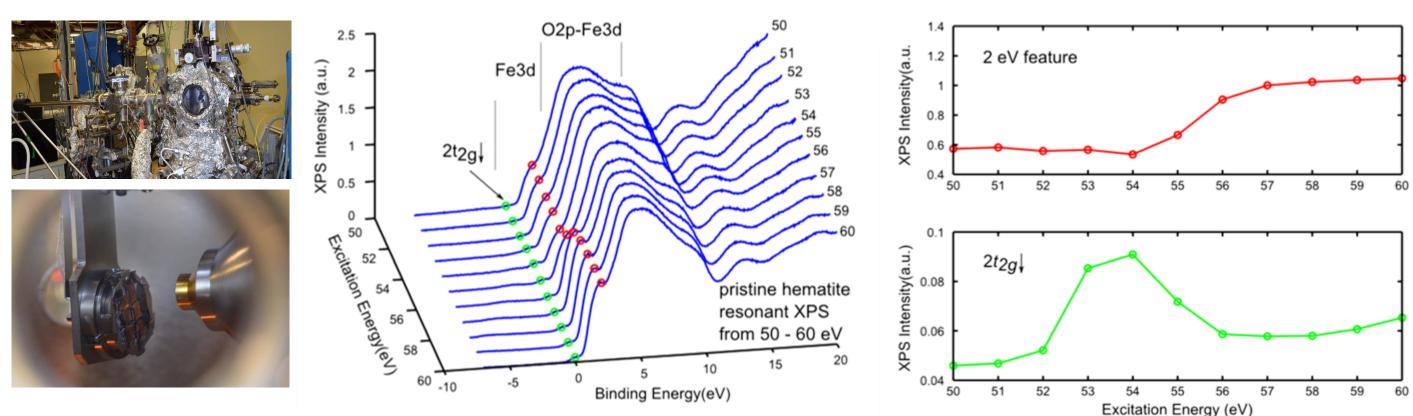


We found that the surface of hematite was oxidized after oxygen plasma treatment, suggesting filling of oxygen vacancies. Combined results from O 1s XPS, formation of iron oxyhydrate is hypothesized.

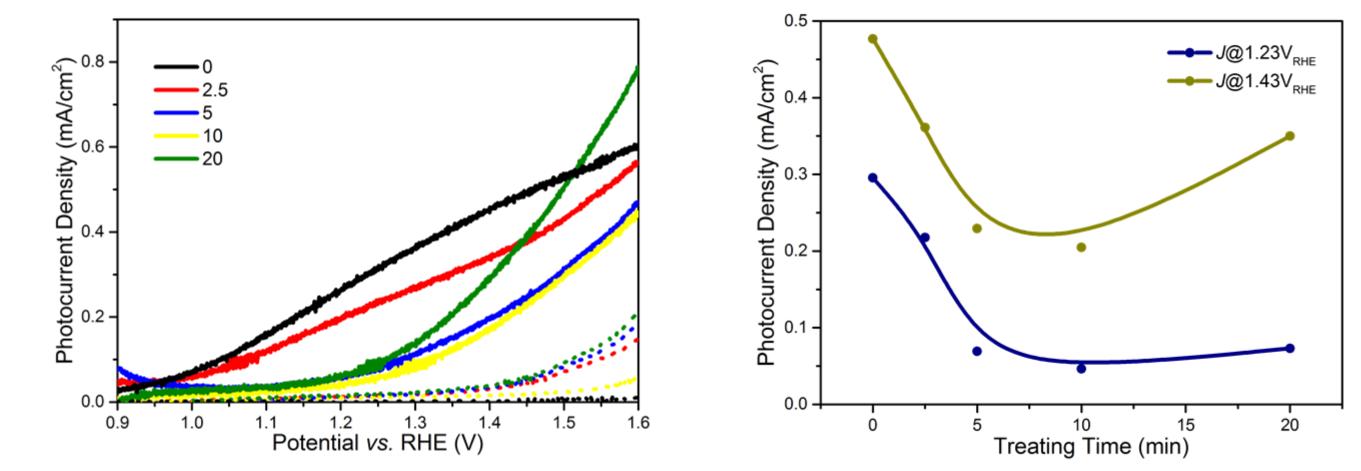
Based on previous theoretical and DFT studies on hematite surface, oxygen vacancy is necessary during water oxidation. It suggests that filling oxygen vacancy after plasma treatment may negatively affect photoelectrochemical properties of hematite photoanode.

- O 1s core level XPS after O₂ plasma treatment
- OH^{-}/O_{x}^{2-} peak (~532 eV) enhanced compared to O^{2-} peak (~ 529.5 eV)
- O^{2-} peak shifted to high binding energy due to OH^{-}/O_{x}^{2-} peak
- Possible reason: Formation of OOH or OH group on surface. Investigation on oxidized state of Fe needed
- Fe 2p core level XPS
- Differences are small between samples thus difficult for analysis
- Other analysis method needed

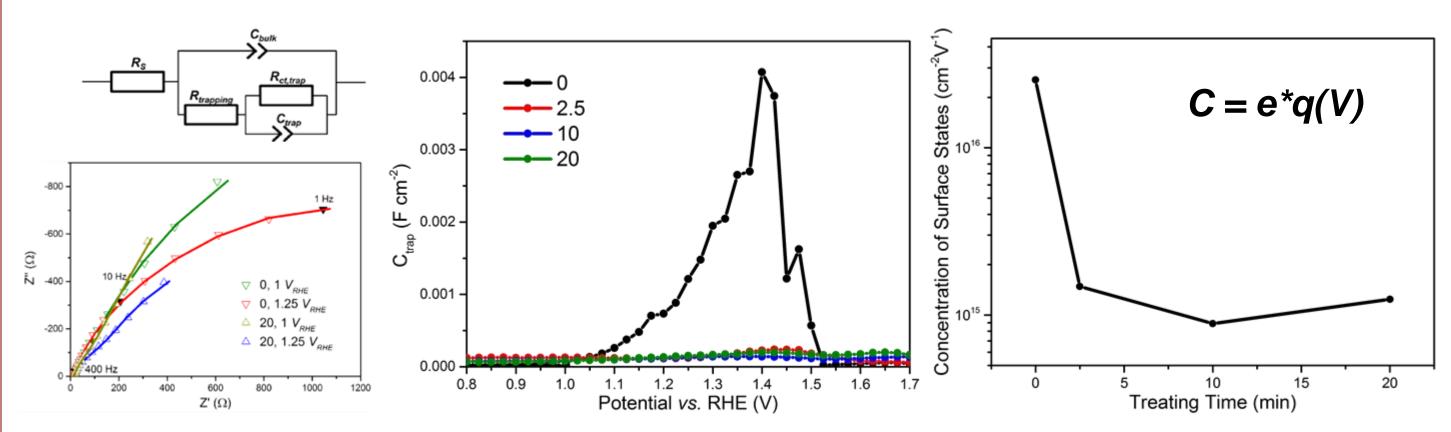
Resonant valence band XPS



Photoelectrochemcial properties and surface states



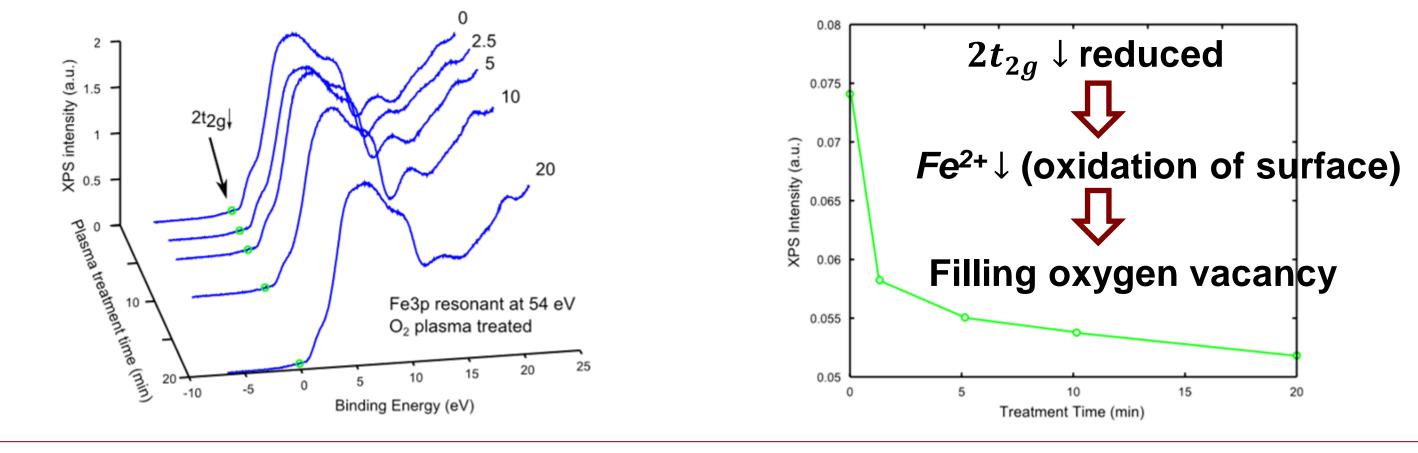
Photocurrent density reduced as expected and affected by treating time



Surface states were studied by impedance spectroscopy under illumination. The capacitance of trapped states suggests the charging/discharging process on surface states. It decreased after oxygen plasma treatment, indicating less photoexicted holes accumulating on surface states for water splitting

2 eV feature is a mixture feature of Fe²⁺ and Fe³⁺, resonant ~ 56

- According to band theory, 0 eV feature is from and **ONLY** from $2t_{2g} \downarrow$ state of Fe²⁺. The undoped hematite contains Fe²⁺ on surface and its resonant energy is around 54 eV
- Resonant energy of 54 eV was selected for best contrast. Intensity of $2t_{2g} \downarrow$ feature vs. treatment time suggests reduced Fe²⁺ concentration after oxygen plasma treatment (oxidation of sample surface)



Reduced concentration of surface states around 1.4 V_{RHE} on surface after treatment

• Combined with surface structure study by O 1s XPS and resonant valence band XPS ($0^{2^{-}}/0H^{-}$ and $2t_{2g} \downarrow$ vs. treating time), an obvious correlation between surface structure, surface states and photoelectrochemical performance was demonstrated

Conclusion

- Oxygen plasma treatment strongly modified hematite surface
- From XPS and resonant valence band XPS, plasma treatment filled oxygen vacancies on hematite surface. Formation of iron oxyhydrate sites were hypothesized.
- Photoelectrochemcial properties of treated hematite samples reduced as expected. Electrochemcial impedance spectroscopy suggest decreasing the concentration of surface states of hematite sample.
- Obvious correlation between surface structure, surface states and photocurrent density was demonstrated.



Fonds national suisse Schweizerischer Nationalfonds Fondo nazionale svizzero Swiss National Science Foundation Schweizerische Eidgenossenschaft Confédération suisse Confederazione Svizzera Confederaziun svizra

Swiss Federal Office of Energy SFOE